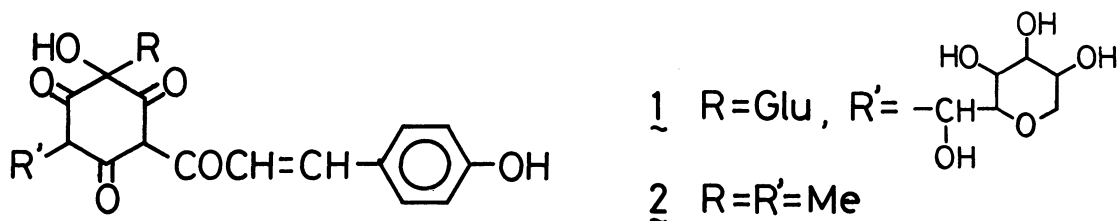


SYNTHESIS OF 2-(p-HYDROXYCINNAMOYL)-4-HYDROXY-4,6-DIMETHYL-
CYCLOHEXANE-1,3,5-TRIONE, AN ANALOG OF SAFFLOMIN A

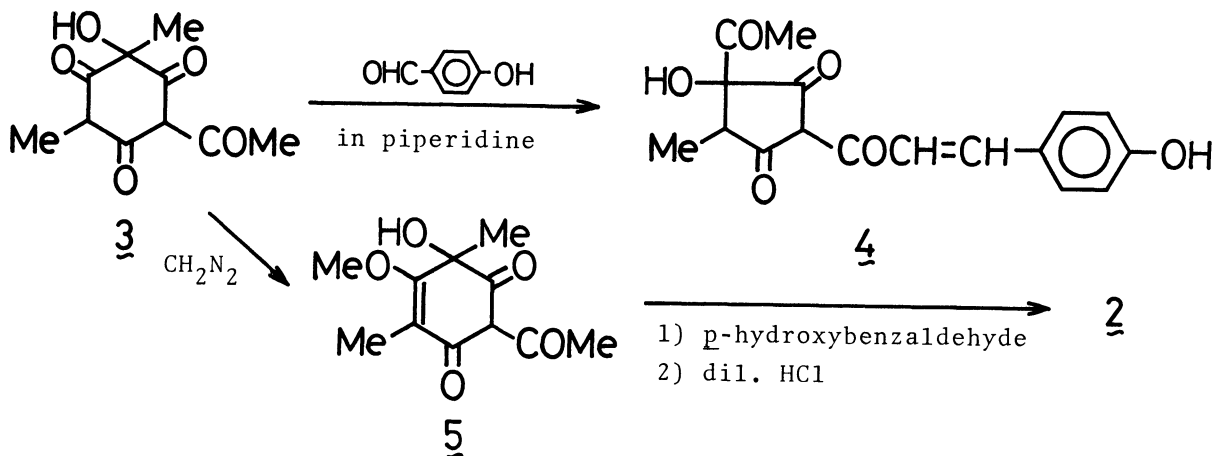
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The synthesis of 2-(p-hydroxycinnamoyl)-4-hydroxy-4,6-dimethyl-
cyclohexane-1,3,5-trione, an analog of safflomin A, was investigated.

In 1981, we have first proposed the structure 1 for safflomin A, a yellow pigment of the flowers of Safflower (*Carthamus tinctorius* L.).¹⁾ However, the synthetic studies of 1 have not been reported yet. In this communication, we wish to report the synthesis of an analog of 1, 2-(p-hydroxycinnamoyl)-4-hydroxy-4,6-dimethylcyclohexane-1,3,5-trione (2), as shown in scheme 1, and the comparison of its behaviours with those of 1.



A mixed solution of 3²⁾ (0.4 g) and p-hydroxybenzaldehyde (0.46 g) in piperidine (2 ml) was stirring at 50 °C for 8 h. The reaction mixture was neutralized with dilute hydrochloric acid and extracted with ethyl acetate. The resulting product was chromatographed on silica gel (Wakogel C-200) using CCl₄-AcOEt-AcOH (60:30:1) to give only 4³⁾ as orange crystals in a 22% yield.



Scheme 1.

The structure of 4 was identified from its $^1\text{H-NMR}$ spectrum.³⁾ It is assumed that the formation of 4 occurred during condensation in piperidine in the same manner as humulone, a bitter constituent of hop, which is easily rearranged to isohumulone with dilute alkali.⁴⁾ Consequently, the protection of 3 against alkali was necessitated. Fortunately, we found that the monomethyl ether (5)⁵⁾ of the enolic form of 3, which was obtained by methylation of 3 with diazomethane, could be used as a protected compound in this condensation. A mixed solution of 5 (0.4 g) and *p*-hydroxybenzaldehyde (1.08 g) in piperidine (6 ml) was stirred at 50 °C for 7 h. The resulting crude crystals were recrystallized from ethyl acetate to afford the monomethyl ether (6)⁶⁾ of 2 in a 40% yield. Deprotection of 6 in methanol containing 10% hydrochloric acid gave 2 quantitatively, and the structure of this target was identified from its spectral data.⁷⁾ Interestingly, silk could be dyed yellow, though cotton could not be colored, with 2 in a similar manner as safflomin A. The electronic spectrum of 2 was also very analogous to that of the natural one (Fig. 1).

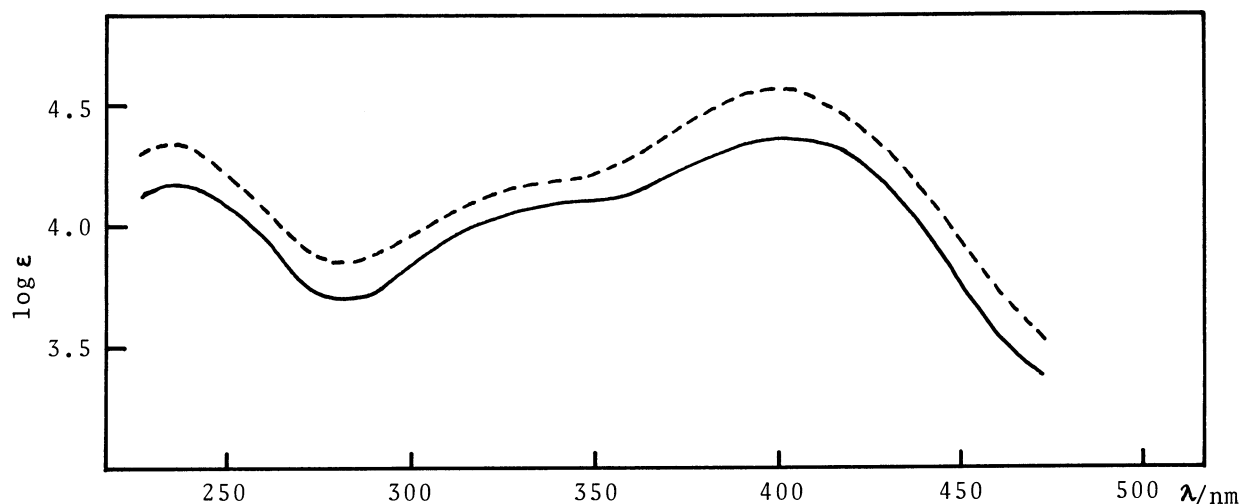


Fig.1. The electronic spectra of safflomin A (—) and 2 (---) in ethanol.

References

- 1) J. Onodera, H. Obara, M. Osone, Y. Maruyama, and S. Sato, *Chem. Lett.*, **1981**, 433.
- 2) T. W. Cambell and G. M. Coppinger, *J. Am. Chem. Soc.*, **73**, 1849 (1951).
- 3) Compound 4; mp 185-186 °C, MS *m/z* 316 (M^+), $\text{UV}\lambda_{\text{max}}$ (EtOH) 244, 352, and 422 nm ($\epsilon = 15300$, 20100, and 13800), $^1\text{H-NMR}$ (DMSO-d_6) δ 1.08 (3H, d, $J=8$ Hz, COCHCH_3), 2.29 (3H, s, COCH_3), 3.12 (1H, q, $J=8$ Hz, COCHCH_3), 6.90 and 7.66 (each 2H, $J=8$ Hz, aromatic protons), 7.50 and 8.04 (each 1H, d, $J=16$ Hz, $-\text{COCH}=\text{CH}-$).
- 4) W. Windish, P. Kolbach, and R. Schleicher, *Wochschr. Brau.*, **44**, 453 (1927); G. A. Howard, *J. Inst. Brewing*, **65**, 417 (1959).
- 5) Compound 5; mp 55-56 °C, MS *m/z* 226 (M^+), $^1\text{H-NMR}$ (CDCl_3) δ 1.57 and 1.85 (each 3H, s, $\text{CH}_3 \times 2$), 2.57 (3H, s, COCH_3), 4.23 (3H, s, OCH_3), 18.71 (1H, s, chelated OH). The yield of this compound from 3 was 61%.
- 6) Compound 6; mp 205-206 °C, MS *m/z* 330 (M^+), $\text{UV}\lambda_{\text{max}}$ (EtOH) 237 and 403 nm ($\epsilon = 12800$ and 37400), $^1\text{H-NMR}$ (DMSO-d_6) δ 1.51 and 1.82 (each 3H, s, $\text{CH}_3 \times 2$), 4.14 (3H, s, OCH_3), 6.90 and 7.57 (each 2H, d, $J=8$ Hz, aromatic protons), 6.80 and 8.02 (each 1H, d, $J=16$ Hz, $-\text{COCH}=\text{CH}-$). $^{13}\text{C-NMR}$ (DMSO-d_6) δ 8.1, 26.8, and 60.4 (each q), 74.9, 104.3, and 111.7 (each s), 111.8, 114.7, and 116.0 (each d), 125.6 (s), 150.6 (d), 160.4, 171.0, 183.9, 190.8, and 195.7 (each s).
- 7) Compound 2; mp 207-208 °C, MS *m/z* 316 (M^+), $^1\text{H-NMR}$ (DMSO-d_6) δ 1.45 and 1.76 (each 3H, s, $\text{CH}_3 \times 2$), 6.87 and 7.57 (each 2H, d, $J=8$ Hz, aromatic protons), 6.81 and 8.02 (each 1H, d, $J=16$ Hz, $-\text{COCH}=\text{CH}-$), 19.15 (1H, s, chelated OH). $^{13}\text{C-NMR}$ (DMSO-d_6) δ 15.1 and 35.7 (each q), 82.4, 111.4, and 112.8 (each s), 123.8 and 126.4 (each d), 133.7 (s), 138.2 and 151.3 (each d), 168.0, 180.9, 191.4, 198.5 and 203.8 (each s).

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